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Processing of reaction-bonded B₄C–SiC composites in a single-mode microwave cavity

Anthony Thuault^{a,*}, Sylvain Marinel^{a,b}, Etienne Savary^{a,c}, Romain Heuguet^a,
Sébastien Saunier^c, Dominique Goeuriot^c, Dinesh Agrawal^b

^aCRISMAT Laboratory UMR 6508 CNRS-ENSICAEN-UCBN, 6 Boulevard du Maréchal Juin, 14050 Caen Cedex, France

^bMaterials Research Institute, Materials Research Laboratory Building, The Pennsylvania State University, University Park, PA 16802, USA

^cDépartement Mécanique et Procédés d'élaboration, Centre des Sciences des Matériaux et des Structures, Ecole Nationale Supérieure des Mines de Saint-Etienne, 42023 Saint-Etienne Cedex 2, France

In this study, the reaction sintering of boron carbide, which consists in doing reactive infiltration of molten silicon throughout a porous sample made of B₄C and carbon graphite was investigated. Thus, it has been shown that a single-mode microwave cavity can be successfully used to produce reaction-bonded B₄C–SiC composite. A specific package, consisting of a SiC based susceptor and a boron nitride based insulating container, was used to heat up the B₄C–Si system using a single-mode microwaves cavity under an Ar–H₂ atmosphere. Pore-free B₄C–SiC composite successfully produced consists of a mixture of B₄C and polygonal shaped β-SiC within a residual silicon matrix. The indentation technique permits to determine mechanical properties of the samples which are compared to those obtained conventionally. It appears that the average hardness ($H \approx 22$ GPa) value is quite constant all along the sample thickness which highlights good homogeneity of the samples obtained. Some aspects of the microstructure are also discussed and compared to those of samples conventionally obtained.

Keywords: A. Microwave processing; C. Mechanical properties; D. Carbides; E. Structural applications

1. Introduction

Boron carbide (B₄C) ceramic is a refractory material with a high melting point (≈ 2723 K), a high hardness, good mechanical properties, a low specific weight (2.52 g cm^{-3}), a high corrosion resistance to chemical agents and a high neutron absorption cross section. B₄C is currently used in many advanced technological application fields [1,2] such as the nuclear industry [3], high-temperature thermoelectricity conversion [4,5] and ballistic protections [6]. As many covalent type carbide ceramics, dense B₄C can be manufactured by a hot-pressing sintering technique at a very high temperature (about 2200 °C), which is a very

costly method limited to the processing of plates or cylindrical samples. Thus, the widespread use of B₄C is restricted by the intrinsic limitations of the hot-pressing technique. Therefore, the Reaction-Bonded Sintering process (RBS-B₄C), which is much more cost-effective, represents an interesting alternative method to produce dense B₄C-based pieces [7]. RBS-B₄C is made by silicon infiltration into a porous preform, made of a B₄C and carbon graphite and shaped by uni-axial pressing. Pure silicon pieces are then put on the top of the preform and the assembly is subsequently heated up to about 1500 °C in order to melt the silicon. The molten silicon infiltrates throughout the boron carbide preform, fills the pores and reacts with graphite. The reaction between silicon and graphite mostly leads to the formation of silicon carbide [8] and, as a result, the composite is mainly made of B₄C grains, SiC grains and there may be some not reacted silicon left [9]. This pressureless and low

*Corresponding author. Tel.: +33 02 31 45 13 77;
fax: +33 02 31 45 13 09.

E-mail addresses: anthony.thuault@ensicaen.fr,
thuault.anthony@yahoo.fr (A. Thuault).

microwaves [16]. Afterwards, the assembly was cooled down to RT during 30 min. The entire heating cycle was conducted under a gas flow made of 95% Ar + 5% H₂. The crystalline phases were identified by X-Ray diffraction (XRD) using Cu K_α radiation (Philips X'Pert diffractometer). The sintered samples were coated in a carbon resin (Struers Polyfast), polished (Struers Tegra-Pol 31) and observed by a scanning electron microscopy (Zeiss Supra 55). Chemical composition was checked by SEM and energy dispersive microscopy (EDAX-EDS). Hardness was determined using a Vickers type micro-durometer with an applying force of 1 kgf and Young's modulus was determined using a nanoindenter (MTS XP), equipped with a Berkovich tip, for a 10,000 nm indentation depth.

3. Results and discussion

Fig. 2A shows a typical cross-section of a sample obtained using the microwave reaction-bonded process. It is clearly shown that the molten silicon has fully infiltrated the porous body. The apparent density is about

2.6 g cm⁻³, which is very close to the expected value (≈ 2.72 g cm³), assuming a complete reaction between graphite and silicon. The sample dimension changes are about 0% and about -5.1% along respectively, the diameter and the height. Compared to the classic sintering process, which leads to shrinkage values ranged between 15% and 20%, it is observed that the reaction-bonded process is a near net shape technique, which is a great advantage compared to the conventional sintering process. The anisotropy of the samples shrinkage observed is not clearly understood yet. It may be due to the fact that the preform has been uni-axially pressed. The SEM microstructure (Fig. 2B) reveals, as expected, a composite microstructure consisting of B₄C and SiC grains, embedded within a silicon matrix. The SiC grains have mostly a polygonal shape, which is very similar to the microstructure reported by Hayun et al. [20] in free carbon added reaction-bonded boron carbide composite. Moreover, the microstructure presents very few pores, which confirms that the infiltration process was successfully completed. SEM observations at various locations of the pellets reveal a quite uniform and homogeneous microstructure at large scale throughout the sample. The XRD pattern performed on a crushed sample is shown in Fig. 3. It appears that the sample seems to be made of B₄C, β -SiC and residual silicon. Hayun et al. [20,21] have deeply studied the microstructure of the reaction-bonded boron carbide made using a conventional method. They used the silicon infiltration process in vacuum (10⁻⁵ Torr) using preform of B₄C with or without carbon addition. They reported that the reaction-bonded boron carbide composite microstructure consists of core-rimmed boron carbide particles, β -SiC and residual Si. They explained that the molten silicon is saturated with free carbon and boron; hence, a reaction due to a secondary equilibrium of the ternary system occurs. The newly formed equilibrium phase, i.e. B₁₂(B,C,Si)₃, precipitates at the interface with the initial boron carbide particles and forms the rim regions. The rim region can be seen using SEM and/or in

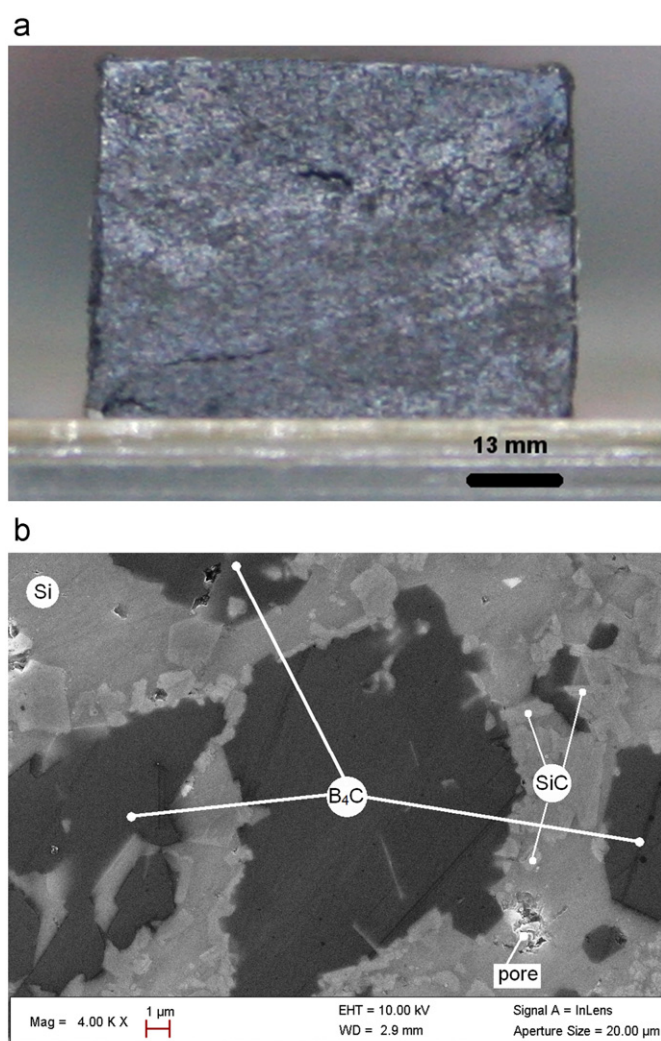


Fig. 2. (a) Cross section of a typical RB B₄C-SiC sample processed using microwaves and (b) its typical microstructure observed using SEM.

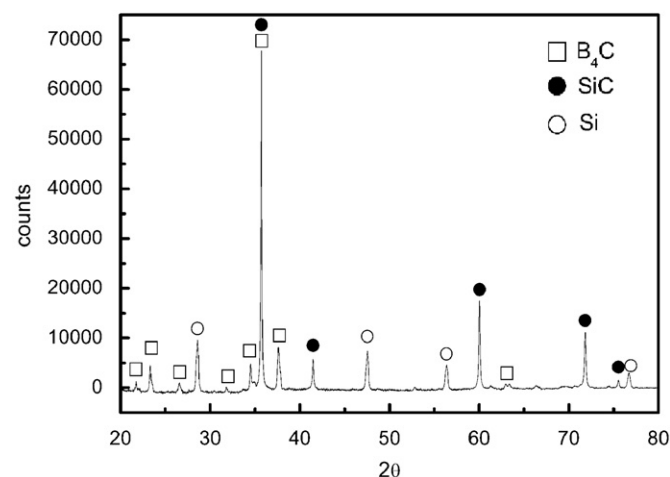


Fig. 3. XRD pattern of a crushed RB B₄C-SiC composite sample.

the XRD pattern, in which a slight split of the ‘B₄C’ peaks is observed. In reaction-bonded sample obtained using microwave sintering, the XRD pattern does not reveal any peak split (Fig. 3). This confirmed that the composite is mainly made of B₄C, β-SiC and residual silicon. It may be assumed that the rapid heating generated by microwaves and the fact that the infiltration process has been done under an Ar–H₂ gas flow at atmospheric pressure may prevent the secondary reaction. Young’s modulus was calculated by nano-indentation along the *z*-axis of the sample and plotted against *z* (depth) in Fig. 4. Measurements were carried out as a matrix 12 × 4 (vertical × horizontal) with an indentation depth of 10,000 nm to obtain large imprints in order to test an area which has a quite homogeneous composition. So, for each depth values, four values of Young’s modulus and hardness were obtained and measurement points were separated by 400 μm. Finally, it appears that the average Young’s modulus is about 309 ± 4 GPa. In addition, the hardness was determined using a Vickers type micro-durometer along the *z*-axis of the sample and plotted against *z* in Fig. 4. Measurements were carried out as a matrix 9 × 4 and the measurement points were separated by 500 μm. Results show an average hardness value of 22 ± 0.5 GPa. It is noted that these mechanical properties are constant throughout the sample, which confirms the microstructure homogeneity and the uniform composition throughout the sample. Because of (i) the obtained microstructure slightly different from the one obtained by conventional process [20,21], since the core-rimmed region has not been observed and because of (ii) the quite large residual silicon phase amount (estimated by calculations to, at least 12.7 vol% if we assumed that all the graphite has reacted but, as the composition was not experimentally determined, it is possible that this amount is underestimated), the average value of Young’s modulus is slightly lower than the one obtained using the conventional process, which is about 360 GPa [22]. However, the average hardness value

obtained in the experiment ($H \approx 22$ GPa) is comparable to the reported value obtained by a conventional reaction-bonded process (about 20–22 GPa) [21]. It can also be noted that those mechanical properties can be improved by adjusting the initial composition to reduce the residual amount of silicon.

4. Conclusion

An original assembly made of a silicon carbide susceptor and a boron nitride based insulating crucible, was designed to produce reaction-bonded boron carbide samples, using a TE₁₀₂ single-mode microwaves cavity. The heating process was successfully carried out, under a flowing Ar–H₂ atmosphere, without noticing any undesirable effects such as arcing or plasma formation. In this study, we have successfully produced a pore free reaction-bonded B₄C–SiC composite, owing to the good silicon infiltration throughout the preform. The final microstructure results in a mixture of B₄C and polygonal shaped β-SiC in a residual silicon matrix. The mechanical properties are comparable to those obtained conventionally and the average hardness value (about 22 GPa) is quite constant all along the sample *z*-axis. In order to produce larger samples, experiments in a multimode microwave system are in progress.

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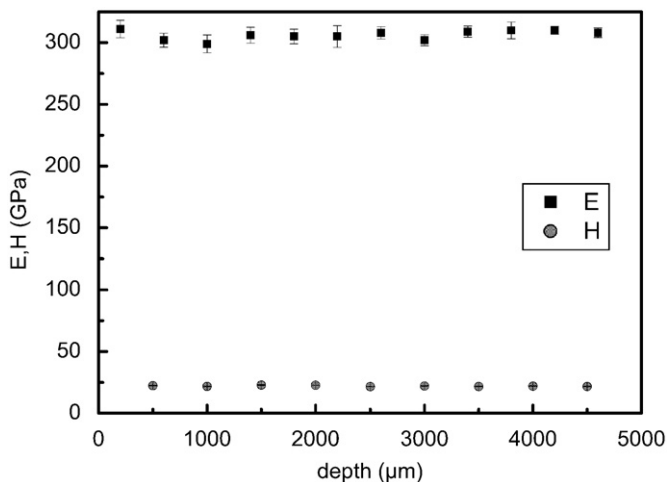


Fig. 4. Representation of Young’s modulus (*E*) and hardness (*H*) values as a function of the depth *z*.

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