

# Hydrogen effects on the plasticity of model fcc and bcc alloys

**David Delafosse** 

Ecole des Mines de Saint-Etienne Centre SMS, UMR CNRS 5307 : Laboratoire Georges Friedel, 158 Cours Fauriel, 42023 Saint-Etienne, France

### **Historic Experiments**





Tritium



Chêne et al., J. Microsc. Spectrosc. Electron., 4 (1979) p. 37-50

H-diffusion in the hydrostatic stress-gradient ahead of a notch



Gruhl, Z. Mtallkd. 75, 11 (1984) p. 819







Step 1 : Implementation & validation in uniaxial tension Step 2 : Application to a cracked bicrystal geometry



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Experimental and analytical modelling of Hydrogen Effects on Tensile Properties

- 1. Elastic binding between mobile solutes and dislocations
- 2. Modelling in the framework of the elastic theory of discrete defects
- 3. H-effects on the tensile properties of fcc single crystals
- **4.** Slip localisation in fcc stainless steels
- **5.** H-effects on the plasticity of bcc FeCr



### **Experimental & Modelling toolbox**

#### Mechanical testing samples

> Model materials : Ni, Ni-16Cr, Fe-15Cr

High purity (less than 25 wt. ppm of impurity)

Single and recrystallised polycrystalline specimens

> Hydrogen charging in autoclave (450°C, P<sub>H2</sub> = 150 to 200 bars, 48 hours)

Control of the hydrogen level and homogeneous concentration

Electrodeposited coating for bcc steels

# > Testing at low temperature in a temperature chamber

Control of the temperature over large testing periods

#### Modelling background

hydrogen

 > Hydrogen effects : coupled elasticitydiffusion equations
 (Larché & Cahn 85, Sofronis & Birnbaum 95, Chateau 2002)
 Based on the elastic theory of crystal defects
 Screening of dislocation pair interactions by

> Plasticity mechanisms : line tension model (De Wit & Koehler 59, Friedel 64)

Simplified (but accurate) analytical description of the elementary plasticity mechanisms in FCC materials



### 1. Elastic binding between mobile H solutes and mobile dislocations



### **1. Dislocation pinning by H solutes**

Dislocation ageing in fcc crystals

1 – Mobile dislocation pinned by solute atmosphere

2 – Instantaneous jump of mobile dislocation between *Forest* obstacles

3 – Ageing of mobile dislocation by diffusing solute atoms during the time required for the thermally activated jump through *Forest* obstacles

4 – Mobile dislocation pinned by solute atmosphere



### **1. Dislocation pinning by H solutes**





# **1. Dislocation pinning by H solutes**



Institut Mines-Télécom

Ni-16Cr :

Saturation for  $t_h > 10^4$ s at -100°C and 10<sup>-2</sup> s<sup>-1</sup>

- ightarrow C<sub>H</sub> = 1500 at. ppm  $\rightarrow$  ~ 13 MPa
- $\rightarrow$  C<sub>H</sub> = 1900 at. ppm  $\rightarrow$  ~ 16 MPa

Ni :

No saturation at -140°C and  $10^{-2} \, s^{-1}$ :

- → Saturation expected for  $t_h > 10^5 s$
- → Effect slightly higher for Ni than for the alloy



### **1. Consequences on Intergranular fracture**



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### 2. Modelling Hydrogen effects on elementary plasticity mechanisms in fcc metals



### **2. Problem statement**

# Hydrogen Enhanced Localised Plasticity mechanism for H-assisted fracture

# based on the shielding of dislocation pair interactions



Influence of H on pile-up density in AISI 310s austenitic stainless steel (Ferreira *et al.* 98)

#### Macroscopic tensile response of H-charged fcc alloys:



### > Macroscopic hardening is observed in (almost) all cases



> often referred to as a "softening"
effect

D.Delafosse – H Embrittlement, NPL Teddington, UK

### 2. Modelling H-dislocation interactions

Solute H (r<sub>H</sub> = 0.53 Å) diffuses in octahedral sites (r<sub>i</sub> = 0.19 Å) → Isotropic distortion of the host lattice : 1<sup>st</sup> order (elastic) effect

Local H flux governed by :

Concentration gradient
 Hydrostatic stress gradient

H diffuses in tension zones :



→ Segregation in the hydrostatic stress field of edge dislocations
 → Relaxation of the hydrostatic stress

#### Steady state distribution





### 2. Modelling H-dislocation interactions

### Screening of pair interactions

Associated stress profile of the same order than that of dislocations (1/r) with opposite sign : screening of dislocations pair interactions

Relative screening effect = independent of the separation distance between dislocations



General expression for the screening of the resolved shear stress between the edge components of any dislocation pair

$$S(T,c) = \frac{S_0}{1+\beta \frac{T}{c}} \text{ with } S_0 = 75\% \quad \beta = \frac{9(1-\nu)RV_M}{2EV^{*2}}$$



#### Chateau JP, Delafosse D, Magnin T. Acta Mater 2002;50:1507.

### 2. Cyclic plasticity in single crystals ...



### ... a (singular) direct consequence of the screening effect



● **7** of the initial cyclic stress amplitude (~10 MPa) : Viscous drag of H atmospheres

**2** Delayed dipoles formation and emergence of the PSB structure :

- **Y** screening of elastic interactions

**3**  $\checkmark$  of  $\tau_s$  (~10%) : increasing of dipoles size :  $\checkmark$  screening of elastic interactions



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# 2. Plasticity mechanisms

General expression for the screening of the resolved shear stress between the edge components of any dislocation pair

$$S(T,c) = \frac{S_0}{1+\beta \frac{T}{c}} \quad \text{with } S_0 = 75\% \qquad \beta = \frac{9(1-\nu)RV_M}{2EV^{*^2}}$$

- ➔ Incorporating S(T,c) in classical plasticity results
  - > Line energy & Line tension (dislocation motion between forest obstacles)
  - > Loop expansion (dislocation multiplication)
  - > Stability of Junctions (hardening)
  - Separation and recombination of partial dislocations (cross-slip of screw dislocations)



Delafosse D. in: Gangloff RP, Somerday BP (Eds.). Gaseous Hydrog. Embrittlement Mater. Energy Technol., vol. 2. Woodhead Publishing Limited; 2012.

### 2. Line Energy & Line Tension

Self-energy  $E(\theta)$  and line tension  $\zeta(\theta)$  :



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### 2. Loop Expansion

Activation and expansion of Franck-Read dislocation loops



$$W^{H} = \frac{3\mu b^{2}L}{4\pi(1-\nu)} \left[ (2-\nu - S(T,c))(0.16 + \ln\left(\frac{L}{r}\right)) \right]$$

Decrease of the force for isotropic expansion of dislocation loops in a line tension model

$$\frac{F^H}{F} = 1 - 0.58 \, \mathbf{S}(\mathbf{T}, \mathbf{c})$$

H in solution decreases the activation stress of Frank-Read sources



### 2. Formation and stability of dislocation junctions









Condition for junction « unzipping »

> Equilibrium of line tensions at the triple point







Delafosse D. in: Gangloff RP, Somerday BP (Eds.). Gaseous Hydrog. Embrittlement Mater. Energy Technol., vol. 2. Woodhead Publishing Limited; 2012.

2. Cross-slip

### Screening of interactions between partial dislocations

Assuming Stacking Fault Energy -  $\Gamma$  = independent of H concentration



Influence of H on the equilibrium shape of an extended node





Ferreira et al. 96

*S*(*T*,*c*) ~ 10%

d Brov R With

Brown & Thölen model (SFE only) :  $\checkmark$  of  $\Gamma$  = 19% With elastic effects :  $\checkmark$  of  $\Gamma$  = 26%

> of the cross-slip probability = core effects + elastic effects

### Decrease of the cross-slip probability by combined elasticity and SFE effects



Delafosse D. in: Gangloff RP, Somerday BP (Eds.). Gaseous Hydrog. Embrittlement Mater. Energy Technol., vol. 2. Woodhead Publishing Limited; 2012.



$$K^{H}(\theta) = \frac{1 - S(T,c) - (v - S(T,c))\cos^{2}\theta}{1 - v}$$

- H favours multiplication by Frank-Read sources :
- Moderate weakening of Lomer junctions for high values of S(T,c), "collapse" of colinear junctions (first stage of hardening)
- Decrease of the cross-slip probability by combined effects on elastic interactions and SFE



$$\frac{F^H}{F} = 1 - 0.58 \, \mathbf{S}(\mathbf{T}, \mathbf{c})$$



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### 3. Tensile behaviour of Hcharged f.c.c. single crystals

(coll. Xavier Faugeas, Université de La Rochelle)



### 3. Tensile behaviour of f.c.c. single crystals



#### Stage I : Planar Glide

- ➔ A unique slip system is activated
- → Low hardening rate, controlled by multiplication and collinear junctions

#### Stage II : Linear hardening

- ➔ Activation of secondary slip systems
- ➔ Hardening rate controlled by the spacing between Geometrically Necessary Boundaries (GNB)

### Stage III : Parabolic hardening

→ Strong influence of the cross-slip mechanism on the decrease of the work hardening rate



# 3. Tensile behaviour of f.c.c. single crystals - Stage I Girardin & Delafosse 2004, Girardin Delafosse & Faugeas 2008



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# 3. Tensile behaviour of f.c.c. single crystals - Stages II-III Huvier, et al., Dislocation 2008, Hong-Kong



Hydrogen content decreases GNBs spacing ( $\lambda$ )

Equiaxed cells seem to disappear (controlled by the cross-slip)

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### 3. Tensile behaviour of f.c.c. single crystals – Stages II-III Huvier, et al., Dislocation 2008, Hong-Kong

3000 300  $<\lambda > (nm)$  TEM  $\theta_{II}$  (MPa) Ni 280 NiH 2500 260 ♦ 135 y = 0.9055xNi16Cr 🔷 240 2000 Ni16Cr-H **001** 220 Ni 1500 200  $\alpha\mu k$ **GNB** 180 1000 160 **IDB** Ni16Cı 140 500 NiH ♦ single glide 120 Ni16Cr-H <d>(nm) AFM  $<\lambda^{-1}>(nm^{-1})$ 0 100 500 1000 1500 2000 0 2500 3000 3500 0 0.0005 0.001 0.0015 0.002 0.00250.003

- Correlation between the inter- band spacing (d) and the inter-wall spacing ( $\lambda$ ) : GNBs act as a periodic distribution of barriers to dislocations motion.

- Hydrogen content decreases the inter- band spacing (d) and the inter-wall spacing ( $\lambda$ ) This effect decreases with a decreasing SFE (not observed for AISI 316L)
- The Hardening rate is a function of an internal length scale :  $\lambda$



# 4. Slip Localisation in austenitic stainless steels

(JM. Olive, I. Aubert, N. Saintier, Université Bordeaux I)



### 4. Slip Localisation in 316L



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### 4. Slip Localisation in 316L

#### Projected SBS versus plastic shear

### Number of dislocations per averaged surface step versus plastic shear



For the same total shear, Hydrogen-charged grains show twice as much projected Slip Band Spacing and twice as many dislocations per surface slip step as hydrogen-free grains



# 5. H-effects on the plasticity of bcc steels

- > Sample preparation for H-charging
- > H-effects on dislocation multiplication
- > H-effects on the dislocation mobility



### 5. Materials



Ferritic stainless steel X14CrMo17 (AISI 430F)

С	S	Si	Mn	Ρ	Cr	Мо	Ni
0.11%	0.35%	0.17%	1.38%	0.025%	16.57%	0.26%	0.38%

Large density of interfaces (matrix/precipitates) maximizing hydrogen trapping



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High purity ferritic binary alloy Fe15Cr

Fe	Cr	С	0	Ν	S
matrix	14.97%	22ppm	12ppm	9ppm	2ppm

High purity material allowing an easier interpretation of the hydrogen – dislocations effects



# **5. Sample preparation for H charging**

Hydrogen diffusion rates and solubilities :



Calculated concentration curves in a α-Fe specimen (Ø5mm) hydrogenated at Co (723K/200bars) desorbing at room temperature  $\partial^2 C$ ЭC C(x;t)/Co  $\frac{\partial D}{\partial t} = D$  $\partial x^2$ After 10s -After 10min After 1h 500 1000 1500 2000 2500 Distance along the radius under the surface (µm)

Hydrogen diffusion rates and solubilities in pure metals at atmospheric pressure:

 $D_{\alpha Fe} (298K) = 9.6 \times 10^{-5} \text{ cm}^2/\text{s}$ ;  $S_{\alpha Fe} (625K) = 1.3 \times 10^{-1}$ wt.ppm  $D_{Ni} (298K) = 5.3 \times 10^{-10} \text{ cm}^2/\text{s}$ ;  $S_{Ni} (625K) = 3.3 \text{ wt.ppm}$  $D_{Cu} (298K) = 6.3 \times 10^{-8} \text{ cm}^2/\text{s}$ ;  $S_{Cu} (625K) = 3.3 \times 10^{-3} \text{ wt.ppm}$ 



Electrodeposited Cu or Ni layer

B.C.C specimen (substrate)

Control of the hydrogen desorption rate by means of Cu or Ni layers electrodeposited prior to gaseous hydrogenation



# **5. Sample preparation for H charging**

• Effects of Cu and Ni layers on total hydrogen concentrations (AISI 430F)



> Desorption kinetics controlled by hydrogen solubility in the electrodeposited metal

> The coatings allow adjusting precisely the hydrogen concentrations in a reproducible way for given hydrogenation conditions (pressure, temperature, time)

> Hydrogen neither desorbs in liquid nitrogen, nor after exposure at room temperature => Subsequent mechanical testing is feasible



# 5. Sample preparation for H charging

• Consequences on hydrogen embrittlement (AISI 430F)

Tensile strength specimen coated and hydrogenated at 723K/200bars/48h and tested at 298K and 203K



Macroscopic embrittlement is an increasing function of H concentration



### 5. H-effects on the uniaxial response at 298K



The order of magnitude of the hydrogen effects on the flow stress is 10MPa/wt.ppm



### 5. H-effects on the uniaxial response at 298K





# 5. Microyielding and dislocation multiplication in pure iron





### 5. Dislocation nucleation in Fe-15Cr













# 5. Dislocation nucleation in Fe-15Cr



Resolved shear stresses corresponding to the pop-in loads in grains A, B and C under different polarizations.

		Slip system	$\tau_{pop-in}$ (GPa)	$ au_{pop-in}/\mu$
Grain A	Anodic Cathodic	(110)[1–11]	6.21 4.98	1/13 1/17
Grain B	Anodic Cathodic	(011)[11-1]	6.75 5.98	1/12 1/14
Grain C	Anodic Cathodic	(10-1)[111]	7.10 6.50	1/12 1/13



- > CRSS values (~µ/12-15) confirm that we are dealing with homogeneous dislocation nucleation
- > They are decreased by 10 to 20% under cathodic polarization



### 5. Thermally activated regime in $\alpha$ -Fe single crystals



T<Tc: Screw segments are straight and move by double kink nucleation and propagation (1 double kink per dislocation line) due to the very high

- Tc<T<Ta: Nucleation of several double kinks per dislocation line. Thanks to the thermal activation, the screw segments bow slightly out and the interations with the forest dislocations are no longer
  - T>Ta: The lattice friction is completely rubbed out by the thermal activation, screw and edge segments have similar mobilities and undergo the line tension model, like in F.C.C materials



# 5. H-effects on the mobility of screw dislocations



Hydrogen in solid solution shifts Ta towards higher temperatures with the order of magnitude of 10K/wt.ppm in the ferritic alloys and decreases the internal stress ( $\approx$  - 5MPa) for hydrogen concentrations of 2-3wt.ppm



### 5. H-effects on the mobility of screw dislocations

**AISI 430F** Reference (0wt.ppm)  $\varepsilon_{\rm p}$ =2% at 133K

Preferrential orientations

Entangled dislocations, curvatures appear

Straight segments



**AISI 430F** Hydrogenated (3,5wt.ppm)  $\epsilon_p=2\%$ à 133K

Preferrencial orientations

Straight dislocations, parallel structures

High density of doublekinks, typical of thermally activated regime



100nm

# 5. Thermal activation in the low temperature regime of b.c.c. crystals

#### $\tau(\mathbf{T},\dot{\varepsilon}) = \tau^*(\mathbf{T},\dot{\varepsilon}) + \tau_i$

- τ\* : Effective stress necessary to overcome obstacles in a thermally activated way (short range interactions: precipitates, lattice friction...)
- $\tau_i$ : Internal stress necessary to overcome long range interactions (elastic interactions: forest dislocations...)

Mobility of screw segments and strain rate:

$$v_{screw} = b \frac{L}{l_c} v_D \frac{b}{l_c} exp\left(\frac{-\Delta H(\tau^*)}{k_B T}\right)$$
  

$$\dot{\gamma} = \dot{\gamma}_0 exp\left(\frac{-\Delta H(\tau^*)}{k_B T}\right)$$

Determination of the double kink nucleation enthalpy:

$$\Delta H(\tau^*) = -\int_{\tau_0^*}^{\tau^*} V(\tau^*) d\tau^*$$

- V : Activation volume
- $au_0^*$  : Effective stress at 0K (Peierls stress)

$$\checkmark \vec{b}$$

Obstacle to be overcome by thermal activation



# 5. Experimental identification of thermal activation parameters



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# 5. Experimental identification of thermal activation parameters



In both alloys, the evolution of the effective stress with the temperature, and the evolution of the activation volume with the effective stress are consistent with known results in pure iron [Smidt, 1969].



# 5. Double kink nucleation and propagation at the microscopic scale



$$\Delta H = 2E_{\rm 1d} + E_{\rm int} - W \quad \text{[Hirth, 1982]}$$

- $\Delta H$ : Double kink nucleation enthalpy
- $E_{1d}$  : Energy of a kink
- $E_{int}$ : Interaction energy of the dislocations segments
- W : Work induced by the applied stress



# 5. Interpretation in terms of H-effects on partial recombination and on double kink nucleation



 $\begin{cases} \dot{\varepsilon} = \dot{\varepsilon}_0 exp\left(\frac{kT}{kT}\right) \\ \vec{\varepsilon}_0 = \frac{\rho_m b^3 v_D L}{l_c^2} \end{cases}$ 

Hydrogen promotes double kink nucleation: Softening effect
 Hydrogen hinders partial recombination (sessile to glissile): Hardening effect
 Shielding effect (elastic interactions): Softening effect



### 5. Summary

- 1. Electrodeposited layers of copper and nickel are provide some control over the H content in tensile specimens of H-charged b.c.c. iron
- 2. Hydrogen effects on tensile properties



3. Hydrogen effect on the athermal temperature Ta (AISI 430F)





### Conclusions

- 1. Solute drag of H atmospheres :
  - Hardening contribution to the flow stress, at room temperature for ordinary strain rates (15-20 MPa for C<sub>H</sub> =1500-2000 at. ppm in pure Ni and binary Ni-16Cr)
- 2. Modelling H-effects at the discrete dislocation scale:
  - > Use of the dimensionless "screening index"
  - > H-effects on line energy and line tension account for several important reaction mechanisms (multiplication, junction formation and stability ...)
  - > The X-slip probability is a mixture of elastic and atomistic properties (SFE and Saddle point configuration energy) and cannot be modelled entirely at this scale
- 3. Tensile tests on single crystals :
  - > Among all the elementary plasticity mechanisms that are liable to Heffects, cross-slip is the one that has the most practical consequences on the tensile response in fcc alloys



### **Conclusions (cont...)**

- 4. Flow localization in austenitic stainless steels:
  - > Even alloys with a low SFE experience an increase of slip planarity & localization
  - > Statistical measurements and detailed crystal plasticity modelling allow for the quantification of this effect
- 5. Experiments are increasingly difficult to carry out in bcc iron, but the control of the H-content via electrodeposited copper yields valuable quantitative results
- H-effects on the flow stress of the order of 20 MPa are obtained with 2-3 ppm of solute H in bcc iron, versus ~ 2000 ppm in fcc nickel or austenitic stainless steels

→ although the basic ingredients are the same, this cannot be rationalized with one single (simple) mechanism – weighting the # contributions requires a closer look at detailed plasticity mechanisms



### **Consequences on fracture** (work in progress)

- > "Reduced ductility" in, e.g.: austenitic stainless steels
  - Effects on slip localization provide a straightforward explanation: shear instability of the ligaments between microvoids. Quantitative models are available for this mechanism
     (e.g. : Liang, Y., Ahn, D. C., Sofronis, P., Mechanics of Materials, 40(2008) p.115)

#### > Brittle Intergranular fracture

Slip localization may participate in the fracture process by increasing the "wedge effect" of slip bands emerging at a grain boundary, but this effect alone is not sufficient to cause IG fracture in ductile materials. Some decrease of GB cohesion is required.

→ need to model the effects of **dynamic hydrogen trapping** at grain boundaries (MD, using EAM and empirical potentials).



### **Consequences on fracture** (work in progress)

- > Brittle Transgranular fracture
  - fcc alloys: Effects on the multiplication, line tension and junction strength will affect both the dynamics of crack-tip shielding and crack blunting, but the net effects are not intuitive because contributions of both signs may appear and need to be quantified
  - The available results may be readily incorporated in Discrete Dislocation Dynamics (DDD) simulations of crack-tip plasticity to quantify H-effects on fracture at the grain scale (Tanguy Pazefindrozaka, Delefered, Acta Mat 56(2008)p 2441)

(Tanguy, Razafindrazaka, Delafosse, Acta Mat.56(2008)p.2441)

In bcc alloys : the thermally activated mobility of screw dislocations is known to be central in the DBT. First quantitative data on H-effects have been obtained (e.g.: shift of the athermal T° towards higher values). Compared to fcc : importance of core effects (as opposed to remote elastic interactions). Need to be investigated further



### Acknowledgements

- Xavier Faugeas (LEMMA, Univ. La Rochelle, France)
- Döme Tanguy (EMSE & Univ Lyon I, France)
- Isabelle Aubert (LMP, Univ. Bordeaux, France)
- Jean-Marc Olive (LMP, Univ. Bordeaux, France & Hydrogenius Center, Univ. Kyushu, Japan)
- Petros Sofronis (Univ. of Illinois)
- Jacques Chêne (Commissariat à l'Energie Atomique, Saclay, France)
- Afrooz Barnoush (NTNU, Norway)

#### Former PhD students @ EMSE :

- Jean-Philippe Chateau
- Gouenou Girardin
- Mialy Razafindrazaka
- Vincent Gaspard
- Flavien Vucko

